

THE STRUCTURES OF CORIOLIN B AND C

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In the previous paper¹⁾, we reported the structure of coriolin (I) which was isolated from a cultured broth of a Basidiomycetes, Coriolus consors. In this paper we wish to describe the structures of coriolin B and C.

Coriolin B (II) m.p. 215-6°, is obtained from the mycelium by extraction with organic solvents. The molecular formula, C₂₃H₃₆O₆ is shown by mass spectrometry and elemental analysis. The IR spectrum shows an ester bond at 1720 cm⁻¹ and hydroxy groups at 3420 and 3450 cm⁻¹. In the NMR spectrum (Fig. 1), a hydrogen on a carbon carrying an esteric oxygen is signaled by a doublet peak at 4.95 δ (J=9.0), and two doublets at 4.43 (J=1.5) and 3.74 δ (J=6.0) suggest the presence of two secondary alcohols.

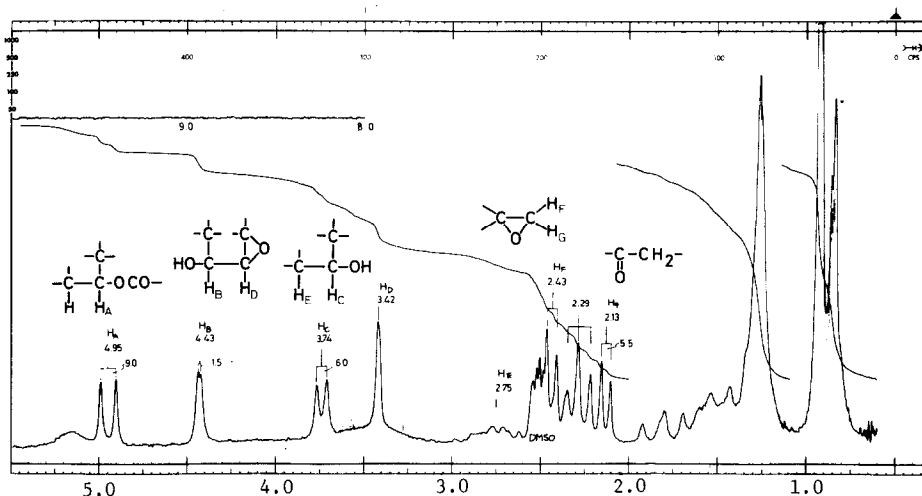
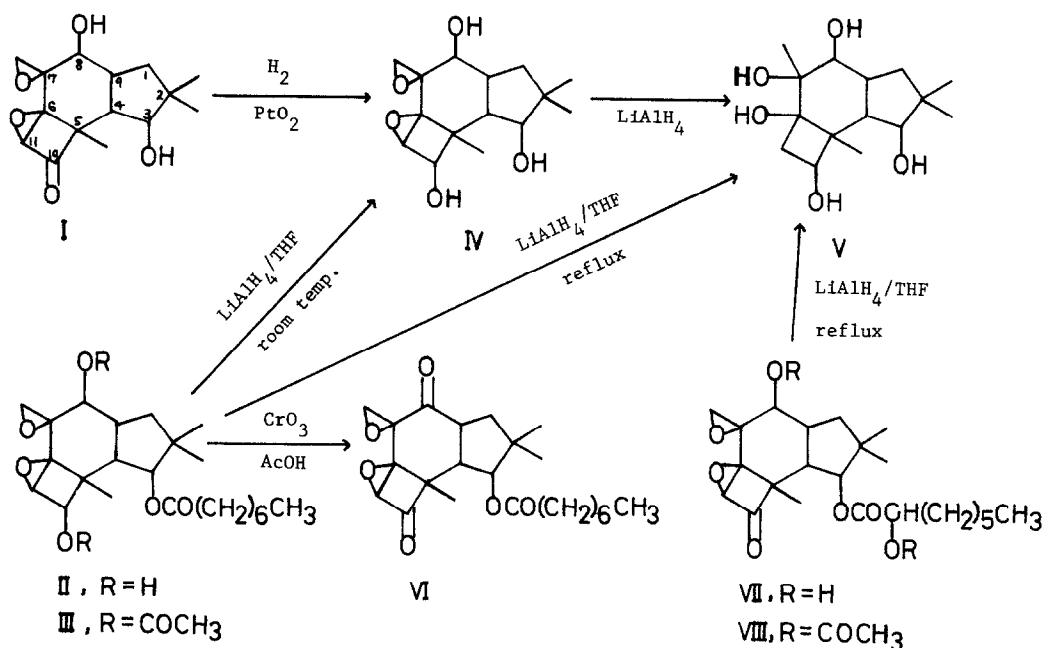


Fig. 1. NMR Spectrum of Coriolin B (100MHz; CDCl₃)



Acetylation of coriolin B with acetic anhydride-pyridine at room temperature gives a diacetate (III) [$\nu_{\max}^{\text{liquid}}$ 1740 cm^{-1} (ester), no hydroxy band; mol. wt. 492 (mass spectroscopy)]. Two carbinol protons (4.43 and 3.74 δ) of the parent compound shift to at 5.59 (J=2.5) and 5.05 δ (J=6.0) in the diacetate, respectively.

Hydrolysis of coriolin B with alkali affords n-caprylic acid which was identified by comparison of IR and mass spectra of the methyl ester with those of the authentic sample.

Reduction of coriolin B with LiAlH₄ in THF at room temperature gives dihydrocoriolin (IV)¹⁾ and n-octyl alcohol (identified by mass spectrometric and gas chromatographic comparisons). With the same reagent, hexahydrocoriolin (V) is obtained under refluxing. From these observations, it is deduced that coriolin B is an ester of dihydrocoriolin and n-caprylic acid.

Oxidation of coriolin B with CrO₃ in acetic acid at room temperature gives a diketone (VI) [C₂₃H₃₂O₆; m.p. 148°; ν_{\max}^{KBr} 1760 (α -epoxy cyclobutanone), 1755 (ester), 1721 cm^{-1} (cyclohexanone), and no hydroxy absorption; mol. wt. 404 (mass spectroscopy)]. This indicates that two secondary hydroxy groups of coriolin B are at C-8 and C-10, and a n-capryl ester at C-3. The position of the ester is confirmed further by comparison of NMR spectra of coriolin B and dihydrocoriolin; that is, C-3 hydrogen of dihydrocoriolin was at 3.4 δ (d, J=9.0) and that of coriolin B shifted to at 4.95 δ (d, J=9.0).

Coriolin C (VII) is another metabolite isolated from the mycelium. The molecular formula $C_{23}H_{34}O_7$ was established by mass spectrometry. The IR spectrum shows the presence of an ester bond at 1740, α -epoxy cyclobutanone function at 1760, and a hydroxy group at 3450 cm^{-1} . The NMR spectrum shows the presence of four methyl groups [0.95 (s), 1.05 (s), 1.15 (s), and 0.95δ (t, $J=6.5$)] and an exocyclic ethylene oxide (2.49 and 3.02δ , $J=6.0$). A singlet peak at 3.47δ indicates the presence of α -epoxy cyclobutanone. A hydrogen on a carbon carrying the ester oxygen is signaled by a doublet peak at 5.09δ ($J=8.0$), and two carbinol protons are present at 3.94.

Reduction of coriolin C with $LiAlH_4$ in THF gives hexahydrocoriolin (V) and n-octan-1,2-diol (identified by IR, MS, and GC comparisons). These observations suggest that coriolin C is an ester of coriolin and α -hydroxy n-caprylic acid.

Acetylation of coriolin C with acetic anhydride-pyridine gives a diacetate (VIII) [$C_{27}H_{38}O_9$; mol. wt. 506 (mass spectroscopy), $\sqrt{\nu_{\max}^{\text{liquid}}}$ 1760, 1740 cm^{-1} , and no hydroxy band]. The NMR spectrum (Fig. 2) shows signals at 5.19 (1H; d, $J=5.5$) and 4.85δ (1H; t, $J=6.0$), indicating the presence of two secondary hydroxy groups in the parent compound. The position of α -hydroxyl-n-capryl ester of coriolin C was determined to be at C-3 by comparison of NMR spectra of coriolin and coriolin C; that is, C-3 hydrogen of coriolin was at 3.47δ (d, $J=9.5$) and that of coriolin C shifted to at 5.09δ (d, $J=8.0$).

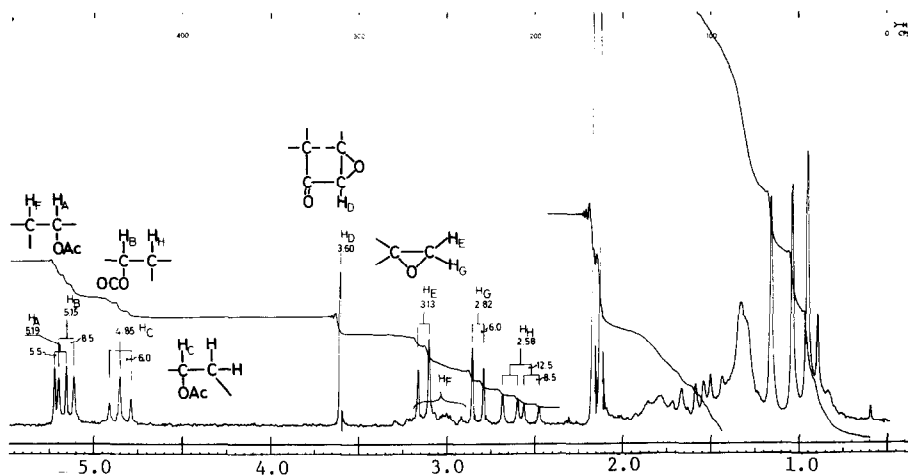


Fig. 2. NMR Spectrum of Coriolin C Diacetate (100MHz; $CDCl_3$)

References

1. S. Takahashi, H. Iinuma, T. Takita, K. Maeda and H. Umezawa, Tetrahedron Letters, No. 53 4663 (1969).